

ON THE KNOWLEDGE OF CUBIC Rb<sub>2</sub>MnF<sub>6</sub>

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Rb<sub>2</sub>MnF<sub>6</sub> has been prepared by the fluorination of the mixture NH<sub>4</sub>MnF<sub>3</sub> + 2RbCl at 350°C. About 500 mg of the sample has been dissolved in 10 ml 40% HF and allowed to crystallize slowly at room temperature. In a few days, light yellow, octahedral single crystals appeared, which were stable when exposed to air. Before the crystal structure analysis, the single crystals have been tempered in fluorine at 320°C for two days. Cubic crystals (Fm3m) with a=8.527(1) Å remained unchanged during treatment.

The structure has been refined from four circle diffractometer data (Philips PW 1100, MoKα, 69 I<sub>0</sub>(hkl), R=5.8 and R<sub>w</sub>=3.3%). The distance d(Mn-F) of 1.782(6) Å (6x) is in accordance with the newly obtained data on K<sub>2</sub>MnF<sub>6</sub>. The Madelung Part of Lattice Energy, MAPLE, for Rb<sub>2</sub>MnF<sub>6</sub> and MnF<sub>4</sub> has been calculated. MAPLE for the latter compound (2573 kcal/mol) is in good agreement with the one from K<sub>2</sub>MnF<sub>6</sub> (2563 kcal/mol). Calculated Effective Coordination Numbers, ECoN, agree with the naif counted values only for Mn<sup>4+</sup> (ECoN=6.0). For Rb<sup>+</sup> and F<sup>-</sup> the ECoN values are 13.2 and 8.9 respectively.